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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.007 Å R factor = 0.088 wR factor = 0.307 Data-to-parameter ratio = 18.4

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1,2-Bis(pyrimidin-2-ylsulfanyl)ethane

The title dithioether compound, $C_{10}H_{10}N_4S_2$, adopts an *anti* conformation with an intramolecular $S \cdots S$ non-bonding distance of 4.419 (2) Å. The dihedral angle between the two pyrimidine rings is 6.8 (2)°.

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Comment

A large number of complexes of dithioether ligands containing *N*-heterocyclic groups have been synthesized and investigated, because of their diverse coordination capabilities and the important properties of their metal complexes (Zheng *et al.*, 2003; Hong *et al.*, 2000). However, crystallographic studies of only a few ligands have been reported. In the present paper, we report the crystal structure of the title compound, (I).



As shown in Fig. 1, the molecule adopts an anti conformation (Goodgame et al., 1999) with a pseudo-torsion angle $C4-S1\cdots S2-C7$ of -174.9 (2)°. The pyrimidine rings (N1/ C1-C3/N2/C4 and N3/C7/N4/C8-C10), with a dihedral angle of 6.8 (2)° to each other, incline to the spacer unit plane (S1 – C5-C6-S2) at angles of 81.8 (4) and 86.9 (4)°, respectively. The S1...S2 non-bonding distance is 4.419 (2) Å. The bond dimensions are within the range reported in a similar 2,4,6-trimethyl-1,3,5-tris(2-pyrimidinylthiocompound, methyl)benzene (Zheng et al., 2002). Both the $S-Csp^2$ and the $S-Csp^3$ bond distances of (I) show little deviation in comparison with those of the analogous compound 1,2-bis-(phenylthio)ethane (Hou et al., 2005), despite different substituents. The molecule of (I) does not possess any crystallographic symmetry, in contrast with the inversion symmetry of several reported alkyl dithioether compounds (Chen et al., 2005; Hou et al., 2005).

In the crystal structure of the binuclear silver(I) nitrate complex with (I) as a chelating and bridging ligand (Zheng *et al.*, 2003), the thioether adopts a *gauche* conformation with an $S \cdot \cdot S$ non-bonding distance of 3.44 (3) Å.

Experimental

Compound (I) was prepared according to the reported procedure (Zheng *et al.*, 2003). Colourless single crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a chloroform solution at room temperature.

Crystal data

 $\begin{array}{l} C_{10}H_{10}N_4S_2\\ M_r = 250.34\\ \text{Monoclinic, } P2_1/c\\ a = 8.0524 \ (16) \ \mathring{A}\\ b = 19.884 \ (4) \ \mathring{A}\\ c = 8.1541 \ (16) \ \mathring{A}\\ \beta = 116.57 \ (3)^\circ\\ V = 1167.7 \ (4) \ \mathring{A}^3\\ Z = 4 \end{array}$

Data collection

Rigaku X-AXIS RAPID IP area-	2663 independent reflections
detector diffractometer	1706 reflections with $I > 2\sigma(I)$
ω scans	$R_{\rm int} = 0.080$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.5^{\circ}$
(ABSCOR ; Higashi, 1995)	$h = -10 \rightarrow 10$
$T_{\min} = 0.725, T_{\max} = 0.936$	$k = -25 \rightarrow 25$
11145 measured reflections	$l = -9 \rightarrow 10$

Refinement

2	
Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.2005P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.088$	+ 0.4011P]
$wR(F^2) = 0.307$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} = 0.046$
2663 reflections	$\Delta \rho_{\rm max} = 1.17 \ {\rm e} \ {\rm A}^{-3}$
145 parameters	$\Delta \rho_{\rm min} = -0.94 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	

 $D_x = 1.424 \text{ Mg m}^{-3}$

Cell parameters from 11145

Mo $K\alpha$ radiation

reflections

T = 293 (2) K

Needle, colourless

 $0.79 \times 0.24 \times 0.16 \ \mathrm{mm}$

 $\theta = 3.0-27.5^{\circ}$ $\mu = 0.43 \text{ mm}^{-1}$

Table 1

Selected geometric parameters (Å, °).

S1 C4	1 750 (4)	S2 C7	1 757 (4)
S1 C5	1.759 (4)	52-C7	1.757 (4)
31-03	1.800 (7)	52-00	1.959 (0)
C4 - S1 - C5	102.6 (2)	C7 - S2 - C6	102.24 (19)
	(-)		
C5-S1-C4-N2	7.7 (4)	C7-S2-C6-C5	88.1 (4)
C4-S1-C5-C6	-83.4(4)	C6-S2-C7-N4	-12.1(4)
S1-C5-C6-S2	-179.0 (2)		

All H atoms were positioned geometrically with $Csp^2 - H = 0.93 \text{ Å}$ and $Csp^3 - H = 0.97 \text{ Å}$; they were constrained to ride on their parent atoms with $U_{iso}(H) = 1.2U_{eq}(C)$. The short C5–C6 distance of 1.425 (7) Å may be an artifact of unresolved disorder, in view of the large atomic displacements. The highest residual electron-density peak is 0.96 Å from atom C5.



Figure 1 ORTEPII (Johnson, 1976) view of (I), showing atomic displacement ellipsoids at the 30% probability level.

Data collection: *RAPID-AUTO* (Rigaku, 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *CrystalStructure* (Rigaku, 2004).

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